

Chemical and Biochemical Engineering

Semester II						
S. No	Course Code	Course Name	L	T	P	C
1	MA109T	<u>Linear Algebra and Differential Equations</u>	3	1.5	0	9
2	ME101C	<u>Engineering Graphics Lab</u>	1	0	3	6
3	ME101L	<u>Hands on Engineering Lab</u>	0	0	3	3
4	CH101T	<u>Introduction to Chemical Engineering</u>	3	0	0	6
5	BB201T	<u>Biomolecules</u>	2	1	0	6
6	CH201T	<u>Organic chemistry</u>	3	0	0	3
7	CC	NSO/NSS/NCC/NCA	0	0	2	2
		Total Credits				35

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Linear Algebra and Differential Equations (3-1.5-0-9)
2	Pre-requisite courses(s)	--
3	Course content	<p>Linear Algebra: Vectors in R^n, notion of linear independence and dependence, linear span of a set of vectors, vector subspaces of R^n, basis of a vector subspace. Systems of linear equations, matrices and Gauss elimination, row space, null space, and column space, rank of a matrix. Determinants and rank of a matrix in terms of determinants. Abstract vector spaces, linear transformations, matrix of a linear transformation, change of basis and similarity, rank-nullity theorem. Inner product spaces, Gram-Schmidt process, orthonormal bases, projections and least squares approximation. Eigenvalues and eigenvectors, characteristic polynomials, eigenvalues of special matrices (orthogonal, unitary, Hermitian, symmetric, skew-symmetric, normal). Algebraic and geometric multiplicity, diagonalization by similarity transformations, spectral theorem for real symmetric matrices, application to quadratic-forms.</p> <p>Differential Equations: Exact equations, integrating factors and Bernoulli equations. Orthogonal trajectories. Lipschitz condition, Picard's theorem, examples on non-uniqueness. Linear differential equations generalities. Linear dependence and Wronskians. Dimensionality of space of solutions, Abel-Liouville formula. Linear ODE's with constant coefficients, the characteristic equations. Cauchy-Euler equations. Method of undetermined coefficients. Method of variation of parameters. Laplace transform generalities. Shifting theorems. Convolution theorem.</p>
4	Texts/References	<ol style="list-style-type: none"> 1. H. Anton, Elementary linear algebra with applications (8th Edition), John Wiley (1995). 2. G. Strang, Linear algebra and its applications (4th Edition), Thomson (2006) 3. S. Kumaresan, Linear algebra - A Geometric approach, Prentice Hall of India (2000) 4. E. Kreyszig, Advanced engineering mathematics (10th Edition), John Wiley (1999) 5. W. E. Boyce and R. DiPrima, Elementary Differential Equations (8th Edition), John Wiley (2005)

Chemical and Biochemical Engineering

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Biomolecules (2-1-0-6)
2	Pre-requisite courses(s)	None
3	Course content	<p>Major classes of biological molecules: Comparison of the alphabets and sources of structural diversity of proteins, nucleic acids, carbohydrates, and lipids.</p> <p>Proteins: Ramachandran plot, evolution of protein structure, structure-function relationships: myoglobin and adaptations in myoglobin structure in deep diving mammals; allostery in hemoglobin; Bohr effect (for pH and carbon dioxide); adult and foetal hemoglobin.</p> <p>Post-translational modifications: special types of covalent bonds found in proteins.</p> <p>Protein folding: Natively folded and natively disordered proteins; miniproteins and peptide toxins; Anfinsen's observations, Levinthal paradox, cooperativity in protein folding, free energy landscape of protein folding and pathways of protein folding, molten globule state, diseases associated with protein folding.</p> <p>Carbohydrates: Sources of structural diversity; structure-function relationship in glycogen and cellulose, Difficulty associated with sequencing of glycans.</p> <p>Lipids: Structure and properties of storage and membrane lipids.</p> <p>Self-assembly of lipids: packing parameter; Biomembrane organization - sidedness and function; membrane bound proteins-structure, properties and function; transport phenomena.</p> <p>Nucleic acids: Historical perspective leading up to the proposition of DNA double helical structure with emphasis on the innovativeness of experimental design; Secondary structure of RNA; chromatin organization.</p> <p>Enzymes: General principles of catalysis; quantitation of enzyme activity and efficiency; Henri-Michaelis-Menten and Briggs-Haldane relationships.</p> <p>Transition state: definition Pauling's intuition and proposal, catalytic antibodies; Catalytic strategies.</p> <p>Isozymes: Haldane relationship between kinetic constants and equilibrium constants; Zymogens.</p> <p>Bioenergetics: basic principles; equilibria and concept of free energy; coupled interconnecting reactions in metabolism; oxidation of carbon fuels, recurring motifs in metabolism. Relevant metabolic pathways may be included to discuss relevant concepts.</p>
4	Texts/References	<ol style="list-style-type: none"> 1. Rodney F Boyer, Concepts in Biochemistry. John Wiley & Sons; 3rd Ed (2 December 2005). 2. Thomas Miilar, Biochemistry Explained: A Practical Guide to Learning Biochemistry CRC Press; 1 edition (30 May 2002). 3. Lubert Stryer et al., Biochemistry.W. H. Freeman; 6th Edition edition (14 July 2006) 4. David L Nelson, and Michael M Cox et al., Lehninger principles of biochemistry WH Freeman; 7th ed. 2017 edition (1 January 2017)

Chemical and Biochemical Engineering

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Engineering Graphics Lab (1-0-3-5)
2	Pre-requisite courses(s)	--
3	Course content	<p>Engineering Graphics with mini drafter: Around half a semester and bit more with following topics to be covered.</p> <ul style="list-style-type: none"> • Introduction to Engineering Graphics • Curves • Projections of Points • Projection of Lines • Projection of Planes • Projections on Auxiliary Planes • Projections of Solids • Sections of Solids • Intersections of Solids <p>Engineering Graphics with 2D Drafting Software: 5 weekly computer laboratory sessions covering above using AutoCAD® as a drafting software, 5th session on Isometric Projections.</p>
4	Texts/References	<ol style="list-style-type: none"> 1. N. D. Bhatt, revised and enlarged by V. M. Panchal and P. R. Ingle, Engineering Drawing, 53rd Edition, 2014, Charotar Publishers, Anand. 2. Warren J. Luzadder and Jon M. Duff, Fundamentals of Engineering Drawing, Prentice-Hall of India. 3. Gopalakrishna K. R., Engineering Drawing Vol. I & II Combined., Subhas Stores, 25th Edition, 2017. 4. Narayana. K. L., and Kannaiah, P. E., Textbook on Engineering Drawing, 2nd Edition, 2013, Scitech Publications, Chennai. 5. Venugopal K. and Prabhu Raja V., Engineering Drawing + AutoCAD, New Age International Publishers, 5th Edition, 2011.

Chemical and Biochemical Engineering

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Hands on Engineering Lab (0-0-3-3)
2	Pre-requisite courses(s)	--
3	Course content	<p>List of Experiments (Mechanical Workshop)</p> <ul style="list-style-type: none"> • To make a Square-fit from the given mild steel pieces (Fitting) • To make a V-fit from the given mild steel pieces (Fitting) • To make a rectangular tray as per required dimensions (Sheet Metal) • To build a transition piece (Sheet Metal) • To make a Butt joint using the given two M.S pieces (Arc welding) • To make a lap joint using the given two M.S pieces (Arc welding) • To build a pipeline using fittings for given flow circuit (Plumbing) <p>List of Experiments (Electrical Workshop)</p> <ul style="list-style-type: none"> • To control one lamp by a one switch with provision for plug socket with switch control (Electrical wiring) • To do stair case wiring (i.e. control of one lamp by two switches fixed at two different places) (Electrical wiring) • Measurement of hot and cold resistance of filament • Improvement of Power Factor • Calibration of Energy meter • Measurement of Power using three ammeter/voltmeter method <p>List of Experiments (Electronics)</p> <ul style="list-style-type: none"> • Understanding breadboard, One-way traffic • Introduction to Arduino and Buzzer • Using Arduino speed measurement of motor/ glowing of LED • Control of water level using Arduino <p>Line follower using Arduino</p>
4	Texts/References	<ol style="list-style-type: none"> 1. Elements of Workshop Technology Vol. 1 (2015), S. K. Hajra Choudhary, A. K. Hajra Choudhary and Nirjhar Roy, Media Promoters and Publishers Pvt. Ltd. 2. W. A. J. Chapman, Workshop Technology, Vol. 1 (2006), Vol 2 (2007), and (1995), CBS Publishers.

Chemical and Biochemical Engineering

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Introduction to Chemical Engineering (3-0-0-6)
2	Pre-requisite courses(s)	Nil
3	Course content	<p>Historical overview of Chemical Engineering: Concepts of unit operations and unit processes, and more recent developments, Features of organized chemical processing- from chemistry to chemical engineering. The Chemical Industry-scope, features & characteristics. and scope. Principles of balancing with examples to illustrate differential and integral balances, lumped and distributed balances. Material balances in simple systems involving physical changes and chemical reactions; systems involving recycle, purge. and bypass.</p> <p>Properties of substances: single component & multicomponent, single and multiphase systems. Use of Compressibility charts, vapour pressure correlations/charts & Psychometric charts. Ideal liquid and gaseous mixtures. Energy balance calculations in simple systems. Introduction to Computer aided calculations-steady state material and energy balances.</p>
4	Texts/References	<ol style="list-style-type: none"> 1. R. M. Felder and R.W. Rousseau, Elementary Principles of Chemical Processes, 3rd ed., John Wiley, New York, 2004. 2. D. M. Himmelblau and J. B. Riggs, Basic Principles and Calculations in Chemical Engineering. 7th ed., Prentice Hall, 2003. 3. B. I. Bhatt and S. M. Vora, Stoichiometry. 4th ed., McGraw Hill, 2004.

Chemical and Biochemical Engineering

Chemical and Biochemical Engineering

1	Title of the course (L-T-P-C)	Organic chemistry (3-0-0-3)
2	Pre-requisite courses(s)	Fundamental concepts and applications of chemistry (CH101)
3	Course content	<p>Reactive Intermediates: An overview of the chemistry of carbenes, nitrenes, radicals, carbocations, carbanions and benzyne. Introduction to substitution, elimination, addition, oxidation, reduction, rearrangement types of reactions</p> <p>Epoxidation named reactions: Jacobsen and Sharpless.</p> <p>Olefin named reactions: Wittig, Julia, Wharton, Peterson, Tebbe.</p> <p>Cross-Coupling named reactions: Buchwald-Hartwig, Negishi, Sonogashira, Suzuki, Wurtz, Ullmann, McMurry, Heck, Stille.</p> <p>Pericyclic reactions: Diels-alder cycloaddition, Ene reaction, Cope rearrangement, Claisen rearrangement (Johnson, Ireland and Eschenmoser).</p> <p>Organic chemistry in industry: Pharmaceuticals, dye, and agrochemicals</p>
4	Texts/References	<ol style="list-style-type: none"> 1. Jerry March and Michael Smith, "Advanced Organic Chemistry", 7th Ed., Wiley, 2015. 2. F. A. Carey and R. J. Sundberg, "Advanced Organic Chemistry, Part A and B", 5th Ed., Springer, 2008. 3. J. Clayden, N. Greeves, and S. Warren, "Organic Chemistry", 2nd Ed., Oxford University Press, 2014. 4. W. Carruthers and I. Coldham, "Modern Methods of Organic Synthesis", 4th Ed., Cambridge University Press, 2015. 5. Laszlo Kurti and Barbara Czako, "Strategic applications of named reactions in organic synthesis", 1st Ed., Elsevier, 2005. 6. R. B. Grossman, "Art of writing reasonable organic reaction mechanisms", 2nd Ed., Springer, 2010. 7. P. Bruice, "Organic Chemistry" 7th Ed., Pearson, 2013. 8. Penny Chaloner, "Organic chemistry: A mechanistic approach, CRC Press; 1st edition, 2014

Chemical and Biochemical Engineering