

## BSMS-Chemistry

Semester V						
S.No	Course Code	Course Name	L	T	P	C
1	CH306T	<u>Coordination and organometallic chemistry</u>	3	0	0	6
2	CH606T	<u>Concepts in organic synthesis</u>	3	0	0	6
3	CH503T	<u>Molecular spectroscopy</u>	3	0	0	6
4	CH401L	<u>Chemistry laboratory II</u>	0	0	3	3
5		Program Elective-II	2	1	0	6
6		Program Elective-III	2	1	0	3
7		Program Elective-IV	2	1	0	3
		<b>Total Credits</b>				<b>33</b>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Coordination and organometallic chemistry (3-0-0-6)</b>
2	<b>Pre-requisite courses(s)</b>	Fundamental concepts and applications of chemistry (CH101) and Transitional metals and coordination chemistry.
3	<b>Course content</b>	<p>Coordination compounds: Valence bond theory, crystal field theory, molecular orbital theory and their applications, inner sphere electron transfer, outer sphere electron transfer, classification of ligands, trans effect, stability constant, Jahn-Teller effect, poly nuclear complexes, reaction of coordination compounds.</p> <p>Bonding and Electronic Spectra: MO theory of transition metal complexes spectroscopic term symbols, selection rules, Orgel diagrams, and charge transfer bands; Magnetism of Coordination Complexes.</p> <p>Organometallic chemistry: General concepts: Types of ligands, soft vs hard ligands. 18e rule and its exceptions, isolobal and isoelectronic analogies. s and <math>\pi</math> bonding, Structure, bonding and reactivity studies of metal carbonyls, nitrosyls, dinitrogen complexes.</p> <p>Organometallic Reactions and Mechanisms: oxidative addition, reductive elimination reactions, organometallic complexes with metal-metal bonds. Metal-ligand Multiple Bonds: Fischer and Schrock type carbene complexes, carbyne complexes, and metal-heteroatom (O/N) multiple bonds</p>
4	<b>Texts/References</b>	<ul style="list-style-type: none"><li>• R. H. Crabtree, The Organometallic Chemistry of the Transition Metals, 6ed, Wiley, 2013.</li><li>• J. Hartwig, Organo-transition Metal Chemistry: From Bonding to Catalysis, University Science Books, 2010.</li><li>• B. D. Gupta and A. J. Elias, Basic Organometallic Chemistry: Concepts, Syntheses and Applications, 2ed, Universities Press, 2013.</li><li>• G. L. Miessler and D. A. Tarr, Inorganic Chemistry, 3ed, Pearson, 2008.</li><li>• B. Douglas, D. McDaniel, and J. Alexander, Concepts and Models of Inorganic Chemistry, 3ed, Wiley, 2010.</li><li>• J. E. Huheey, E. A. Keiter, and R. L. Keiter, Inorganic Chemistry Principles of Structure and Reactivity, 4ed, Pearson Education, 2006</li><li>• Inorganic Chemistry. D. F. Shriver, and P. W. Atkins. 3rd Edn. Oxford University, Oxford, 1999.</li><li>• S. F. A. Kettle, Physical Inorganic Chemistry – A Coordination Chemistry Approach, Springer, 1996.</li></ul>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Concepts in organic synthesis (3-0-0-6)</b>
2	<b>Pre-requisite courses(s)</b>	Fundamental concepts and applications of chemistry (CH101) and organic reactions and reagents
3	<b>Course content</b>	Synthesis, reactions, mechanisms, and selectivity involving the following-alkenes, alkynes, arenes, alcohols, phenols, aldehydes, ketones, carboxylic acids and their derivatives, halides, nitro compounds and amines. Use of compounds of Mg, Li, Cu, B and Si in organic synthesis. Concepts in multistep synthesis- retrosynthetic analysis, disconnections, synthons, synthetic equivalents, linear and convergent synthesis, reactivity umpolung, selectivity, protection and deprotection of functional groups, Asymmetric synthesis: Chiral auxiliaries, methods of asymmetric induction – substrate, reagent and catalyst-controlled reactions; enantiomeric and diastereomeric excess; enantio-discrimination. Resolution – optical and kinetic.
4	<b>Texts/References</b>	<ul style="list-style-type: none"><li>Clayden, J., Greeves, N., Warren, S., Wothers, S. <i>Organic Chemistry</i>, Oxford University Press, <b>2001</b>.</li><li>Carruthers, W., Coldham, I. <i>Some Modern Methods of Organic Synthesis</i>, Cambridge University Press, <b>2004</b>.</li><li>Smith, M. B. and March, J. <i>Advanced Organic Chemistry</i>, Wiley Interscience, <b>2007</b>.</li><li>Carey, F. A., Sundberg, R. J. <i>Advanced Organic Chemistry, Part A and B</i>, Springer, <b>2007</b>.</li><li>Smith, M. B. <i>Organic Synthesis</i>, McGraw-Hill, <b>2001</b>.</li><li>Warren, S. <i>Organic Synthesis: The Disconnection Approach</i>, Wiley, <b>1983</b>.</li><li>G. S. Zweifel and M. H. Nantz, <i>Modern Organic Synthesis-An Introduction</i>, W. H. Freeman and Company, 2006</li></ul>

## BSMS-Chemistry

1	<b>Title of the course</b> (L-T-P-C)	<b>Molecular spectroscopy</b> <b>(3-0-0-6)</b>
2	<b>Pre-requisite courses(s)</b>	Fundamental concepts and applications of chemistry (CH101)
3	<b>Course content</b>	<p>Introduction to spectral energy domains and measurement of spectra, Implications of discrete energy levels, Population of States – Boltzman Distribution, Interaction of radiation with matter, origin of linewidths in molecular spectra, Transition dipole moment and Fermi's Golden Rule, Einsteins Coefficients, Lasers and Masers.</p> <p>Rotational (Microwave) spectroscopy, Molecular vibrations - Infrared spectroscopy, Normal mode analysis, Raman Scattering, Selection Rules from Group Theory, Molecular electronic spectra, Photophysical processes, Non-Linear Spectroscopy, Nuclear Magnetic Resonance, Relaxation times, FT-NMR, spin-spin coupling, ESR, Nuclear Quadrupolar Resonance.</p>
4	<b>Texts/References</b>	<ol style="list-style-type: none"><li>1. J. L. McHale, Molecular Spectroscopy, Pearson Education, 1999.</li><li>2. M. Hollas, Modern Spectroscopy, Wiley; 4th edition, 2004.</li><li>3. F. A. Cotton, Chemical Applications of Group Theory, 3rd edition, Wiley-Interscience, 1990.</li><li>4. D. C. Harris, M. D. Bertolucci, Symmetry and Spectroscopy, Dover, 1990.</li><li>5. C. M. Banwell, E. M. McCash, Fundamentals of Molecular Spectroscopy, Tata McGraw Hill, 1983</li><li>6. G. M. Barrow, Molecular Spectroscopy, McGraw Hill, 1962</li><li>7. J. I. Steinfeld, Molecules and Radiation: An Introduction to Modern Molecular Spectroscopy, 2nd edition, Dover, 2005.</li><li>8. J. D. Graybeal, Molecular Spectroscopy, McGraw Hill 1993.</li><li>9. D. A. McQuarrie and J. D. Simon, Physical Chemistry - a molecular approach, Viva Books Pvt. Ltd. 1998.</li></ol>

## BSMS-Chemistry

1	<b>Title of the course (L-T-P-C)</b>	<b>Chemistry laboratory II</b> <b>0-0-3-3</b>
2	<b>Pre-requisite courses(s)</b>	--
3	<b>Course content</b>	Inorganic chemistry: Complex material analyses: minerals/ alloys. Quantitative estimations using conductometry and spectrophotometry. Organic chemistry: Qualitative analysis of organic compounds. Chemical separation of binary mixtures and their qualitative analysis, Synthesis of organic compounds and chromatography Physical chemistry: Phase equilibria, viscosity and molecular weight of polymers, surface tension, reaction kinetics (rates, order of reaction, influence of ionic strength), use of thermocouples, transition temperature determinations.
4	<b>Texts/References</b>	<ul style="list-style-type: none"><li>• G. Svehla and B. Sivasankar, “Vogel's qualitative inorganic analysis”, Pearson Education India, 7th Ed, 2023.</li><li>• G. Mendham, R. C. Denney, J.D. Barnes, M. Thomas, B. Sivasankar, “Vogel's quantitative chemical analysis” Pearson education India, 6th Ed. 2009.</li><li>• A. J. Elias, “A collection of interesting general chemistry experiments” Sangam Books Ltd. First Ed. 2002.</li><li>• B. Viswanathan, and P. S. Raghavan, Practical Physical Chemistry, Viva Books, 1st Ed., 2010</li><li>• A. M. Halpern, and G. C. McBane, Experimental Physical Chemistry: A Laboratory TextBook, 3rd Edition, W. H. Freeman, 2006</li><li>• B. S. Furniss, A. J. Hannaford, P.W.G. Smith, A.R. Tatchell, “Vogel's textbook of practical organic chemistry” Pearson education India, 5th Ed. 2003.</li><li>• In-house laboratory manual with the experimental procedures and relevant literature.</li></ul>