

# Chemical and Biochemical Engineering

Semester II						
S.N o	Course Code	Course Name	L	T	P	C
1	MA109T	<u>Linear Algebra and Differential Equations</u>	3	1.5	0	9
2	ME101C	<u>Engineering Graphics Lab</u>	1	0	3	6
3	ME101L	<u>Hands on Engineering Lab</u>	0	0	3	3
4	CL101T	<u>Introduction to Chemical Engineering</u>	3	0	0	6
5	BB201T	<u>Biomolecules</u>	2	1	0	6
6	CH201T	<u>Organic chemistry</u>	3	0	0	3
7	CC	NSO/NSS/NCC/NCA	0	0	2	2
		Total Credits				35

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1	<b>Title of the course (L-T-P-C)</b>	<b>Linear Algebra and Differential Equations (3-1.5-0-9)</b>
2	<b>Pre-requisite courses(s)</b>	--
3	<b>Course content</b>	<p><b>Linear Algebra:</b> Vectors in <math>\mathbb{R}^n</math>, notion of linear independence and dependence, linear span of a set of vectors, vector subspaces of <math>\mathbb{R}^n</math>, basis of a vector subspace. Systems of linear equations, matrices and Gauss elimination, row space, null space, and column space, rank of a matrix. Determinants and rank of a matrix in terms of determinants. Abstract vector spaces, linear transformations, matrix of a linear transformation, change of basis and similarity, rank-nullity theorem. Inner product spaces, Gram-Schmidt process, orthonormal bases, projections and least squares approximation. Eigenvalues and eigenvectors, characteristic polynomials, eigenvalues of special matrices (orthogonal, unitary, Hermitian, symmetric, skew-symmetric, normal). Algebraic and geometric multiplicity, diagonalization by similarity transformations, spectral theorem for real symmetric matrices, application to quadratic-forms.</p> <p><b>Differential Equations:</b> Exact equations, integrating factors and Bernoulli equations. Orthogonal trajectories. Lipschitz condition, Picard's theorem, examples on non-uniqueness. Linear differential equations generalities. Linear dependence and Wronskians. Dimensionality of space of solutions, Abel-Liouville formula. Linear ODE's with constant coefficients, the characteristic equations. Cauchy-Euler equations. Method of undetermined coefficients. Method of variation of parameters. Laplace transform generalities. Shifting theorems. Convolution theorem.</p>
4	<b>Texts/References</b>	<ol style="list-style-type: none"><li>1. H. Anton, Elementary linear algebra with applications (8th Edition), John Wiley (1995).</li><li>2. G. Strang, Linear algebra and its applications (4th Edition), Thomson (2006)</li><li>3. S. Kumaresan, Linear algebra - A Geometric approach, Prentice Hall of India (2000)</li><li>4. E. Kreyszig, Advanced engineering mathematics (10th Edition), John Wiley (1999)</li><li>5. W. E. Boyce and R. DiPrima, Elementary Differential Equations (8<sup>th</sup> Edition), John Wiley (2005)</li></ol>

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1	Title of the course (L-T-P-C)	Biomolecules (2-1-0-6)
2	Pre-requisite courses(s)	None
3	Course content	<p><b>Major classes of biological molecules:</b> Comparison of the alphabets and sources of structural diversity of proteins, nucleic acids, carbohydrates, and lipids.</p> <p><b>Proteins:</b> Ramachandran plot, evolution of protein structure, structure-function relationships: myoglobin and adaptations in myoglobin structure in deep diving mammals; allostery in hemoglobin; Bohr effect (for pH and carbon dioxide); adult and foetal hemoglobin.</p> <p><b>Post-translational modifications:</b> special types of covalent bonds found in proteins.</p> <p><b>Protein folding:</b> Natively folded and natively disordered proteins; miniproteins and peptide toxins; Anfinsen's observations, Levinthal paradox, cooperativity in protein folding, free energy landscape of protein folding and pathways of protein folding, molten globule state, diseases associated with protein folding.</p> <p><b>Carbohydrates:</b> Sources of structural diversity; structure-function relationship in glycogen and cellulose, Difficulty associated with sequencing of glycans.</p> <p><b>Lipids:</b> Structure and properties of storage and membrane lipids.</p> <p><b>Self-assembly of lipids:</b> packing parameter; Biomembrane organization - sidedness and function; membrane bound proteins-structure, properties and function; transport phenomena.</p> <p><b>Nucleic acids:</b> Historical perspective leading up to the proposition of DNA double helical structure with emphasis on the innovativeness of experimental design; Secondary structure of RNA; chromatin organization.</p> <p><b>Enzymes:</b> General principles of catalysis; quantitation of enzyme activity and efficiency; Henri-Michaelis-Menten and Briggs-Haldane relationships.</p> <p><b>Transition state:</b> definition Pauling's intuition and proposal, catalytic antibodies; Catalytic strategies.</p> <p><b>Isozymes:</b> Haldane relationship between kinetic constants and equilibrium constants; Zymogens.</p> <p><b>Bioenergetics:</b> basic principles; equilibria and concept of free energy; coupled interconnecting reactions in metabolism; oxidation of carbon fuels, recurring motifs in metabolism. Relevant metabolic pathways may be included to discuss relevant concepts.</p>
4	Texts/References	<ol style="list-style-type: none"> <li>1. Rodney F Boyer, Concepts in Biochemistry. John Wiley &amp; Sons; 3rd Ed (2 December 2005).</li> <li>2. Thomas Miilar, Biochemistry Explained: A Practical Guide to Learning Biochemistry CRC Press; 1 edition (30 May 2002).</li> <li>3. Lubert Stryer et al., Biochemistry. W. H. Freeman; 6th Edition edition (14 July 2006)</li> <li>4. David L Nelson, and Michael M Cox et al., Lehninger principles of biochemistry WH Freeman; 7th ed. 2017 edition (1 January 2017)</li> </ol>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Engineering Graphics Lab (1-0-3-5)</b>
2	<b>Pre-requisite courses(s)</b>	--
3	<b>Course content</b>	<p>Engineering Graphics with mini drafter: Around half a semester and bit more with following topics to be covered.</p> <ul style="list-style-type: none"><li>• Introduction to Engineering Graphics</li><li>• Curves</li><li>• Projections of Points</li><li>• Projection of Lines</li><li>• Projection of Planes</li><li>• Projections on Auxiliary Planes</li><li>• Projections of Solids</li><li>• Sections of Solids</li><li>• Intersections of Solids</li></ul> <p>Engineering Graphics with 2D Drafting Software: 5 weekly computer laboratory sessions covering above using AutoCAD® as a drafting software, 5th session on Isometric Projections.</p>
4	<b>Texts/References</b>	<ol style="list-style-type: none"><li>1. N. D. Bhatt, revised and enlarged by V. M. Panchal and P. R. Ingle, Engineering Drawing, 53rd Edition, 2014, Charotar Publishers, Anand.</li><li>2. Warren J. Luzadder and Jon M. Duff, Fundamentals of Engineering Drawing, Prentice-Hall of India.</li><li>3. Gopalakrishna K. R., Engineering Drawing Vol. I &amp; II Combined., Subhas Stores, 25th Edition, 2017.</li><li>4. Narayana. K. L., and Kannaiah, P. E., Textbook on Engineering Drawing, 2nd Edition, 2013, Scitech Publications, Chennai.</li><li>5. Venugopal K. and Prabhu Raja V., Engineering Drawing + AutoCAD, New Age International Publishers, 5th Edition, 2011.</li></ol>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Hands on Engineering Lab (0-0-3-3)</b>
2	<b>Pre-requisite courses(s)</b>	--
3	<b>Course content</b>	<p><b>List of Experiments (Mechanical Workshop)</b></p> <ul style="list-style-type: none"> <li>• To make a Square-fit from the given mid steel pieces (Fitting)</li> <li>• To make a V-fit from the given mid steel pieces (Fitting)</li> <li>• To make a rectangular tray as per required dimensions (Sheet Metal)</li> <li>• To build a transition piece (Sheet Metal)</li> <li>• To make a Butt joint using the given two M.S pieces (Arc welding)</li> <li>• To make a lap joint using the given two M.S pieces (Arc welding)</li> <li>• To build a pipeline using fittings for given flow circuit (Plumbing)</li> </ul> <p><b>List of Experiments (Electrical Workshop)</b></p> <ul style="list-style-type: none"> <li>• To control one lamp by a one switch with provision for plug socket with switch control (Electrical wiring)</li> <li>• To do stair case wiring (i.e. control of one lamp by two switches fixed at two different places) (Electrical wiring)</li> <li>• Measurement of hot and cold resistance of filament</li> <li>• Improvement of Power Factor</li> <li>• Calibration of Energy meter</li> <li>• Measurement of Power using three ammeter/voltmeter method</li> </ul> <p><b>List of Experiments (Electronics)</b></p> <ul style="list-style-type: none"> <li>• Understanding breadboard, One-way traffic</li> <li>• Introduction to Arduino and Buzzer</li> <li>• Using Arduino speed measurement of motor/ glowing of LED</li> <li>• Control of water level using Arduino</li> </ul> <p>Line follower using Arduino</p>
4	<b>Texts/References</b>	<ol style="list-style-type: none"> <li>1. Elements of Workshop Technology Vol. 1 (2015), S. K. Hajra Choudhary, A. K. Hajra Choudhary and Nirjhar Roy, Media Promoters and Publishers Pvt. Ltd.</li> <li>2. W. A. J. Chapman, Workshop Technology, Vol. 1 (2006), Vol 2 (2007), and (1995), CBS Publishers.</li> </ol>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Introduction to Chemical Engineering (3-0-0-6)</b>
2	<b>Pre-requisite courses(s)</b>	<b>Nill</b>
3	<b>Course content</b>	<p>Historical overview of Chemical Engineering: Concepts of unit operations and unit processes, and more recent developments, Features of organized chemical processing- from chemistry to chemical engineering. The Chemical Industry-scope, features &amp; characteristics. and scope. Principles of balancing with examples to illustrate differential and integral balances, lumped and distributed balances. Material balances in simple systems involving physical changes and chemical reactions; systems involving recycle, purge, and bypass.</p> <p>Properties of substances: single component &amp; multicomponent, single and multiphase systems. Use of Compressibility charts, vapour pressure correlations/charts &amp; Psychometric charts. Ideal liquid and gaseous mixtures. Energy balance calculations in simple systems. Introduction to Computer aided calculations-steady state material and energy balances.</p>
4	<b>Texts/References</b>	<ol style="list-style-type: none"><li>1. R. M. Felder and R.W. Rousseau, Elementary Principles of Chemical Processes, 3rd ed., John Wiley, New York, 2004.</li><li>2. D. M. Himmelblau and J. B. Riggs, Basic Principles and Calculations in Chemical Engineering. 7th ed., Prentice Hall, 2003.</li><li>3. B. I. Bhatt and S. M. Vora, Stoichiometry. 4th ed., McGraw Hill, 2004.</li></ol>

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1	<b>Title of the course (L-T-P-C)</b>	<b>Organic chemistry (3-0-0-3)</b>
2	<b>Pre-requisite courses(s)</b>	Fundamental concepts and applications of chemistry (CH101)
3	<b>Course content</b>	<b>Reactive Intermediates:</b> An overview of the chemistry of carbenes, nitrenes, radicals, carbocations, carbanions and benzyne. Introduction to substitution, elimination, addition, oxidation, reduction, rearrangement types of reactions <b>Epoxidation named reactions:</b> Jacobsen and Sharpless. <b>Olefination named reactions:</b> Wittig, Julia, Wharton, Peterson, Tebbe. <b>Cross-Coupling named reactions:</b> Buchwald-Hartwig, Negishi, Sonogashira, Suzuki, Wurtz, Ullmann, McMurry, Heck, Stille. <b>Pericyclic reactions:</b> Diels-alder cycloaddition, Ene reaction, Cope rearrangement, Claisen rearrangement (Johnson, Ireland and Eschenmoser). <b>Organic chemistry in industry:</b> Pharmaceuticals, dye, and agrochemicals
4	<b>Texts/References</b>	1. Jerry March and Michael Smith, “Advanced Organic Chemistry”, 7 <sup>th</sup> Ed., Wiley, 2015. 2. F. A. Carey and R. J. Sundberg, “Advanced Organic Chemistry, Part A and B”, 5 <sup>th</sup> Ed., Springer, 2008. 3. J. Clayden, N. Greeves, and S. Warren, “Organic Chemistry”, 2nd Ed., Oxford University Press, 2014. 4. W. Carruthers and I. Coldham, “Modern Methods of Organic Synthesis”, 4 <sup>th</sup> Ed., Cambridge University Press, 2015. 5. Laszlo Kurti and Barbara Czako, “Strategic applications of named reactions in organic synthesis”, 1 <sup>st</sup> Ed., Elsevier, 2005. 6. R. B. Grossman, “Art of writing reasonable organic reaction mechanisms”, 2 <sup>nd</sup> Ed., Springer, 2010. 7. P. Bruice, “Organic Chemistry” 7 <sup>th</sup> Ed., Pearson, 2013. 8. Penny Chaloner, “Organic chemistry: A mechanistic approach, CRC Press; 1st edition, 2014

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